
Communications to the Editor

PLURACIDOMYCIN A₂, A NEW
CARBAPENEM BEARING A SULFINIC
ACID, AND OTHER MINOR
PLURACIDOMYCINS

Sir:

Streptomyces pluracidomyceticus PA-41746 (FERM BP-174) was found to produce additional β -lactam components besides the previously reported pluracidomycins (PLM) A, B and C¹⁾. In particular, fermentation at 38°C resulted in an increase of the minor components with a decrease of PLM C, while the yields of PLM A and PLM B remained almost unchanged. Reported here are the isolation and structure elucidation of these minor pluracidomycins.

The pluracidomycins were isolated from the fermentation filtrate as previously reported using phase-transfer extraction, ion-exchange chromatography on QAE-Sephadex and gel filtration of the respective fractions to obtain fractions A, B, C and D (in the order of decreasing acidity).

Fraction A containing the main component PLM A₁ (formerly PLM A), a new minor component PLM A₂, and a considerable amount of PLM B, was separated by column chromatography on MCI GEL CHP20P (Mitsubishi Chem. Ind.) using 15% NaCl solution. Fraction B contained a small amount of PLM A₂ and the second major component PLM B and was separated with a CHP20P column, as was fraction A. The PLM A₂ portions were combined, desalted by gel filtration on Biogel P-2 and freeze-dried.

Fraction C was further fractionated on CHP20P using 10%, 5% and 2% NaCl solutions (stepwise) into fractions C₁, C₂, C₃ and C₄. Fraction C₁ gave PLM C₁ (formerly PLM C). The active principles of fractions C₂ and C₃ were assumed to be new compounds according to HPLC data but those of fraction C₄ were identified as MM 4550 and MM 17880²⁾. Fraction C₂ was rechromatographed on CHP20P using 5% NaCl, desalted and freeze-dried. When the minor component was purified by HPLC on Nucleosil ₅C₈ (Macherey-Nagel) with 0.1 M phosphate buffer (pH 7.0), desalted and freeze-dried, a new component PLM C₂ was obtained

in extremely poor yield. Further purification of fraction C₃ on CHP20P column using 3% NaCl solution, desalting and freeze-drying gave a new component PLM C₃.

Fraction D had three active components, MM 4550, MM 17880 and a new substance according to the HPLC data. In chromatography with a CHP20P column, the new substance, PLM D, was eluted by 10% NaCl solution, while the two known antibiotics were eluted by water. The PLM D fractions were combined, desalted and freeze-dried.

The yields of the new components, isolated from about 500 liters of filtrate as respective sodium salts, were approximately 40 mg of PLM A₂, 3 mg of PLM C₂, 15 mg of PLM C₃ and 100 mg of PLM D. The ¹H NMR spectra in D₂O of these pluracidomycins are shown in Fig. 1, and the relative mobilities on paper electrophoresis are listed in Table 1.

The ¹H NMR spectrum of PLM A₂ is closely related to that of PLM A₁; some differences were observed in the chemical shifts of the protons at C-4 and C-5, but not in the total proton number and the proton-proton coupling constants. The ¹³C NMR data (Table 2), in accordance with ¹H NMR, clearly indicated that PLM A₂ has nine carbon atoms as has PLM A₁ and differs from PLM A₁ only in the substituent at C-3. As shown in Table 1, PLM A₂ revealed a mobility equal to that of PLM A₁ and therefore the substituent at C-3 should be an acidic function lacking the carbon atom, like that of PLM A₁. Since the UV absorption ($\lambda_{\text{max}}^{\text{H}_2\text{O}}$ 273 nm) is different from those of PLM A₁, B and C₁¹⁾, the substituent should be a new type of carbapenem antibiotic. Conjecture about its function was based on comparison of the IR spectra (Fig. 2). The spectrum of PLM A₁ has a strong band (1200~1270 cm⁻¹) attributable to the S=O stretching of the -SO₃⁻ and the -O-SO₃⁻ functions. In the spectrum of PLM A₂, however, the corresponding band is narrowed due to lack of the absorption at 1200 cm⁻¹, and two new bands appear at 1020~1060 cm⁻¹ and at 965 cm⁻¹, which may be assignable to the S=O and S-O stretching of a sulfinate³⁾. The molecular formula of PLM A₂ was determined to be

Fig. 1. ^1H NMR spectra of pluracidomycins A₂ (1), C₂ (2), C₃ (3) and D (4) at 200 MHz. Chemical shifts are shown in δ (ppm) from external TMS (in D₂O).

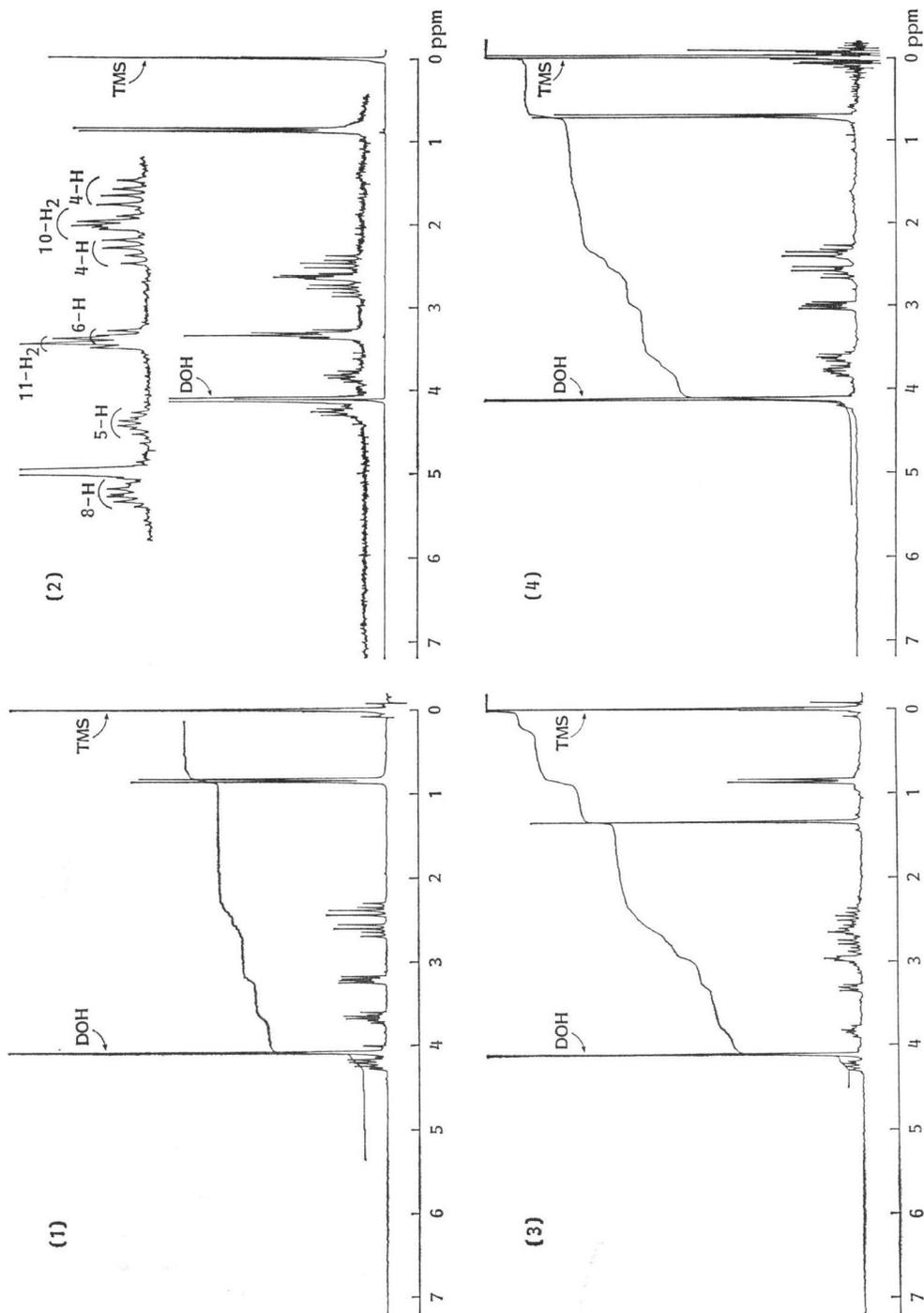


Table 1. Paper electrophoresis (1/30 M phosphate buffer (pH 7.0); 12 V/cm; 1.5 hours).

Antibiotic	Relative mobility
PLM A ₂	3.0
PLM C ₂	1.9
PLM C ₃	1.9
PLM D	2.3
PLM A ₁	3.0
Epithienamycin A	1.0

C₉H₈O₉NS₂Na₃ by mass spectrometry (SIMS); the MS gave m/z 408 [M+H]⁺ and m/z 430 [M+Na]⁺ whereas that of PLM A₁ gave m/z 424 [M+H]⁺ and m/z 446 [M+Na]⁺. In order to confirm the 3-sulfino structure, PLM A₂ was treated with sodium *m*-chloroperbenzoate in phosphate buffer (pH 7.0). The oxidation proceeded smoothly at room temperature and, as expected, gave PLM A₁ in almost quantitative yield.

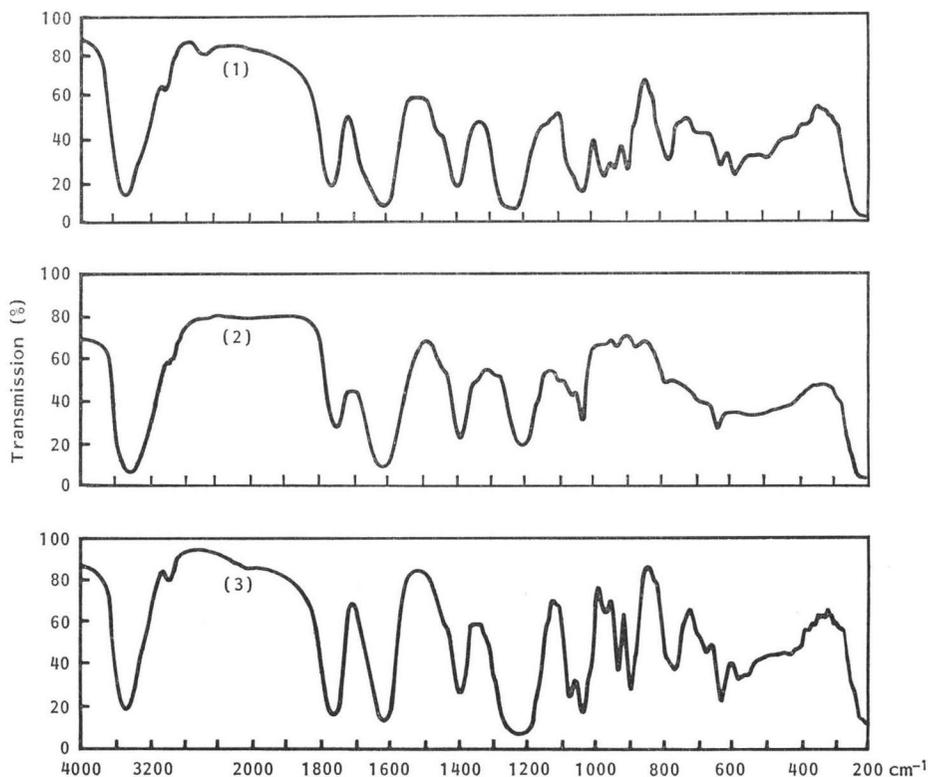
A few naturally occurring sulfinic acids related to hypotaaurine have been isolated from animal sources; for example, hypotaurocyamine

Table 2. ¹³C NMR of pluracidomycins in D₂O (δ ppm).

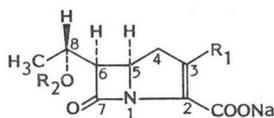
C	PLM A ₁	PLM A ₂	PLM D
2	139.2 (s)	129 (s)	139.8 (s)
3	129.0 (s)	152.0 (s)	127.7 (s)
4	32.8 (t)	27.0 (t)	32.4 (t)
5	55.1 (d)	54.5 (d)	55.0 (d)
6	58.7 (d)	57.8 (d)	60.4 (d)
7	168.7 (s)	167.4 (s)	169.0 (s)
8	73.7 (d)	73.9 (d)	64.4 (d)
2-COONa	177.9 (s)	178.4 (s)	179.4 (s)
8-CH ₃	19.4 (q)	19.4 (q)	21.7 (q)

H₂N $\overset{\text{O}}{\parallel}$ NCNHCH₂CH₂SOH has been isolated from marine annelids⁴⁾. PLM A₂ seems to be the first sulfinic acid identified as a microbial metabolite.

PLM C₂ is presumably a member of the sulfated olivanic acid group bearing a sulfinyl function at C-3 based on its UV ($\lambda_{\text{max}}^{\text{H}_2\text{O}}$ 287 nm) and ¹H

Fig. 2. IR spectra of pluracidomycins A₂(1), D(2) and A₁(3). (KBr).

Pluracidomycin (PLM)



	R ₁	R ₂
PLM A ₂	$\begin{array}{c} \text{O} \\ \uparrow \\ \text{S}-\text{ONa} \end{array}$	SO ₃ Na
PLM C ₂	$\begin{array}{c} \text{O} \\ \uparrow \\ \text{S}-\text{CH}_2\text{CH}_2\text{OH} \end{array}$	SO ₃ Na
PLM C ₃	$\begin{array}{c} \text{O} \\ \uparrow \\ \text{S}-\text{CH}_2\text{CH}_2\text{NHAc} \end{array}$	SO ₃ Na
PLM D	SO ₃ Na	H
PLM A ₁	SO ₃ Na	SO ₃ Na
PLM B	$\begin{array}{c} \text{O} \\ \uparrow \\ \text{S}-\text{CH}_2\text{COONa} \end{array}$	SO ₃ Na
PLM C ₁	$\begin{array}{c} \text{O} \\ \uparrow \\ \text{S}-\text{CH} \begin{array}{l} \text{OH} \\ \text{OH} \end{array} \end{array}$	SO ₃ Na
MM4550	$\begin{array}{c} \text{O} \\ \uparrow \\ \text{S}-\text{CH}=\text{CHNHAc} \end{array}$	SO ₃ Na
MM17880	S-CH ₂ CH ₂ NHAc	SO ₃ Na

NMR spectra. Its ¹H NMR showed two significant methylene proton signals near 3.26 ppm (2 H, m) and near 3.95 ppm (2 H, m) attributable

to $\begin{array}{c} \text{O} \\ \uparrow \\ -\text{SCH}_2\text{CH}_2\text{X} \end{array}$. The X group should be a hydroxyl group, because of the downfield shift of the signals of one methylene group compared with that of asparenomycin B⁵⁾ and the absence of the CH signals due to the X group. Though the sequences of CH were confirmed by spin-spin decoupling experiments, further evidence to confirm the structure could not be obtained due to the limited amount of this component. PLM C₂ seems to be a precursor of PLM B.

PLM C₃ was easily deduced to be the sulfoxide of MM 17880 by comparison of its ¹H NMR with those of MM 17880 and asparenomycin B. The downfield shift of the proton at C-5 (slightly) and of SCH₂ (remarkably) clearly shows the sulfinyl structure of the side-chain⁵⁾. The molecular formula was confirmed to be C₁₃H₁₆O₆N₂S₂Na₂ by SIMS: *m/z* 455 [M+H]⁺, 477 [M+Na]⁺. Also, the sulfinyl function was easily reduced by titanium trichloride in acetate buffer⁵⁾, and the product was identified as MM 17880 by

HPLC.

PLM D has a chromophore (UV λ_{max}^{0.0} 232, 267 nm) similar to that of PLM A₁ but its paper electrophoresis suggested that presence of only two acidic functions (Table 1). Comparison of the ¹H and ¹³C NMR of PLM D and PLM A₁ indicated that PLM D is a compound corresponding to the desulfated PLM A₁ at C-8. The IR spectrum of PLM D, in agreement with the above assumption, lacked the absorption bands at 1250 cm⁻¹, 935 cm⁻¹ and 895 cm⁻¹ which are commonly observed in the spectra of sulfated olivanic acids. The mass spectrum (SIMS) supported the molecular formula, C₉H₉O₇NSNa₂; *m/z* 322 [M+H]⁺, 344 [M+Na]⁺.

The carbapenem antibiotics are generally labile in aqueous solutions, however, PLM D is fairly stable against hydrolysis at neutral condition. The stability seems to be comparable to that of C-19393 H₂⁶⁾ (carpetimycin B).

These minor pluracidomycins also showed broad antibacterial and β-lactamase inhibitory activities, as will be reported elsewhere.

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